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Chapter 9 Chemical Names And Formulas Assessment Answers ...

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126 Chapter 9 Chemical Formulas Example 9.3 What is the empirical formula for a compound having the composition: 49.0% C, 2.7% H and 48.3% Cl? 1. One hundred amu of the compound contains 49.0 amu C, 2.7 amu H and 48.3 amu Cl. 2. Mol values are $49.0/12.0 = 4.08$ for C, $2.7/1.0 = 2.7$ for H, and $48.3/35.5 = 1.36$ for Cl. 3.

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regular metals include Group 1 (except H) and 2 and aluminum. When naming a compound that starts with a regular metal, name the first element and add -ide to the second element (except for polyatomic ions).

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Chapter 9 Practice Test - Naming and Writing Chemical Formulas

Chapter 9 Chemical Names and Formulas85 Names and Formulas for Bases (page 273) 6. A base is a compound that produces ____ when dissolved in water. 7. How are bases named? SECTION 9.5 THE LAWS GOVERNING FORMULAS AND NAMES (pages 274-279) This section uses data to demonstrate that a compound obeys the law of definite proportions.

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The name of a base is the name of the cation followed by the name of the anion (hydroxide). The formula of a base is written by showing the number of hydroxide ions needed to balance the positive charge on the cation. After reading Lesson 9.4, answer the following questions. Names and Formulas of Acids hydrogen1.

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