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statement is true; if the statement is
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Assessment (Part A)Match the equation in Column A with its description in Column B. Column A Column B 1.

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1- Gases consist of large numbers of tiny particles that are far apart relative to their size. 2- Collisions between gas particles and between particles and container walls are elastic collisions. 3- Gas particles are in continuous, rapid, random motion and possess kinetic energy. 4- No forces of attraction between gas particles.

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Understanding Main Ideas (Part A) Match the equation in Column A with its

description in Column B. 2. 3. 4. Column A = h/mv AE E higher-energy orbit lower-energy orbit a. b. c. d. Column B Relates the wavelength, frequency, and speed of an electromagnetic wave Describes the energy change of an electron undergoing an orbit transition

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